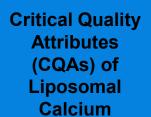




## Summary of Characterizations Performed on Liposomal Calcium



Identification and quantification of lipid species



#### API

Quantification of the encapsulated, un-encapsulated, and total API

#### **Liposomal Products**

- Morphology and structure (lamellarity)
- Mean particle size and size distribution
- Surface charge (zeta potential)

## S

#### **Stability**

- Physical stability (fusion and aggregation)
- Chemical stability (degradation of lipids and API)



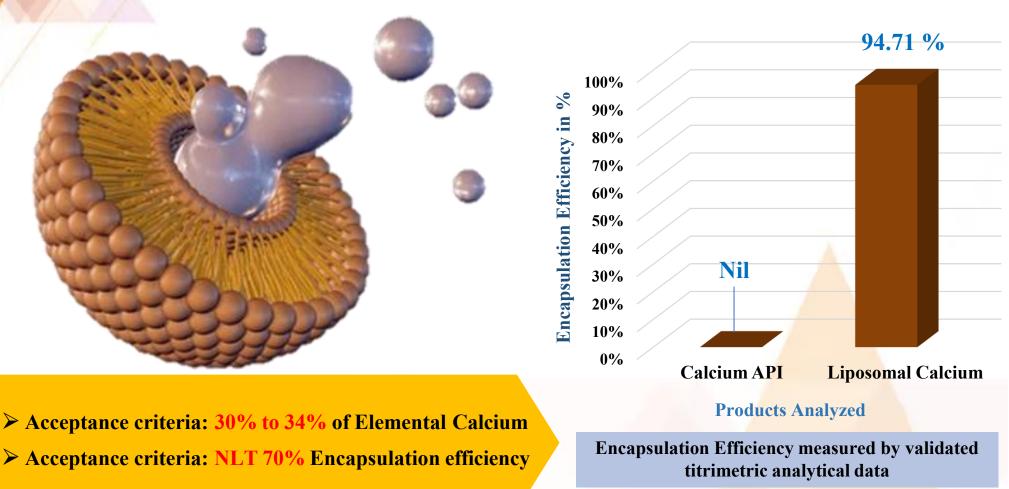
In vitro release kinetics of the encapsulated API

- 1. Encapsulation efficiency of Liposomal Calcium
- 2. Analysis of particle size and uniformity of Liposomal Calcium using DLS
- 3. Behavior of Liposomal Calcium particles in liquid medium using DLS Zeta-sizer
- 4. FTIR analysis of Liposomal Calcium
- 5. Elemental Analysis of Liposomal Calcium
- 6. Morphology analysis of Calcium Liposomes using SEM
- 7. Analysis of Calcium leakage from Liposomes
- 8. Stability analysis of Liposomes at 105° C temperatures
- 9. Endothermic study of Liposomal Calcium using DSC analysis
- 10. Mineral Loading Capacity
- 11. Product Uniformity Mesh Size vs EE%





### 1. Encapsulation Efficiency of 29.02% Liposomal Calcium



- Liposomal encapsulation ensures 94.71% efficiency, significantly surpassing the minimum requirement of 70%.
- Efficient encapsulation minimizes mineral loss, improving bioavailability and therapeutic efficacy.
- > Offers protection against oxidation and gastrointestinal irritation, common with conventional calcium forms.



## 2. Dynamic Light Scattering Analysis of Liposomal Calcium

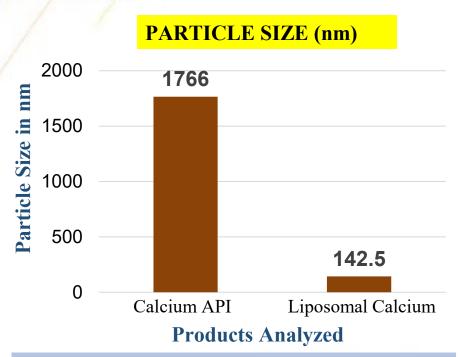


Figure 1 – Chart showing the particle size of Calcium API with Liposomal Calcium



- ➤ Particle Size : < 220 nm
- Polydispersity Index : <1</p>

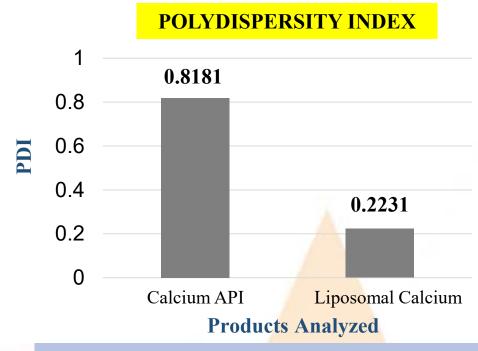


Figure 2 – Polydispersity Index (PDI) of Liposomal Calcium in solution

- Nanosized, uniform particles offer greater colloidal stability and improved shelf life.
- Smaller particles (particle size: 142.5 and PDI 0.2231) support increased mucosal permeability and cellular uptake.
- DLS characterization confirms high formulation control and batch-to-batch reproducibility.



## 3a. Behavior of Liposomal Calcium

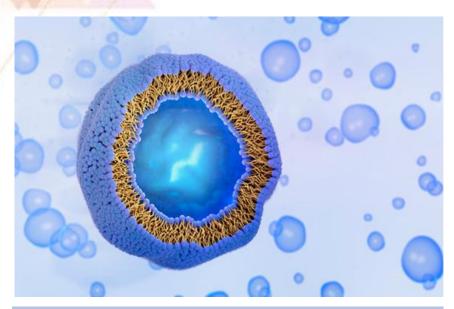


Figure 1 – A figure showing the balance of attractive and repelling forces which determine how particles behave in a medium.



> Zeta Potential : < -30 mV

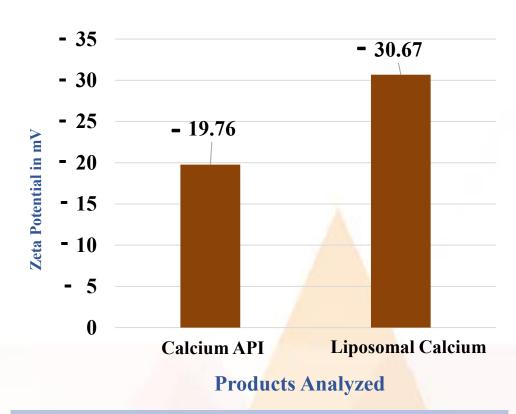
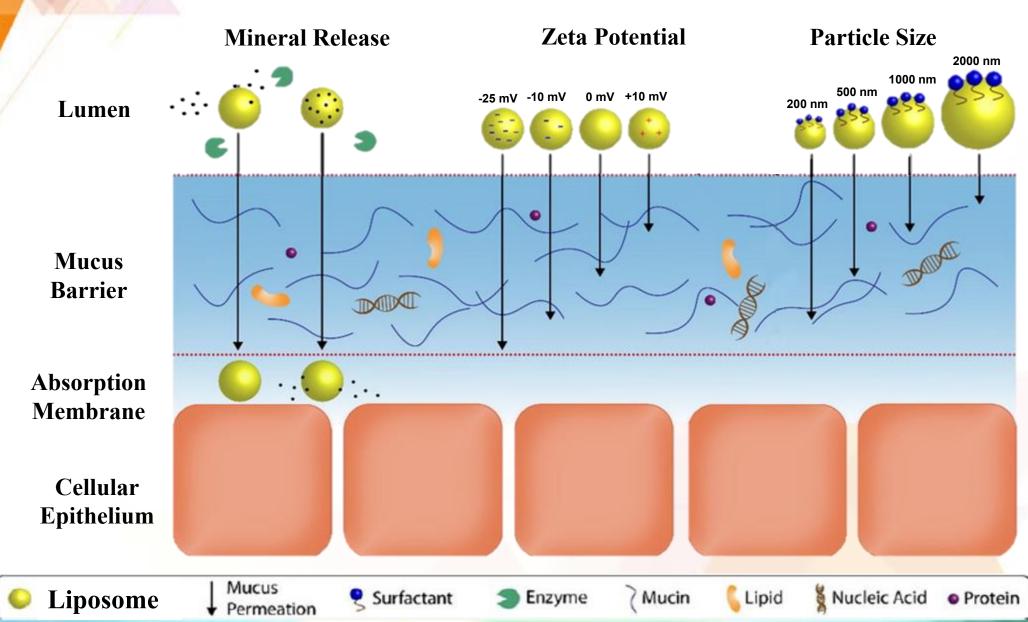


Figure 2 – Chart comparing the Zeta potential of Calcium API and Liposomal Calcium showing Calcium in Liposomal form is stable and unlikely to agglomerate in solution.

- ➤ Liposomal Calcium shows high zeta potential (-30.67 mV) → excellent colloidal stability.
- $\triangleright$  Prevents particle aggregation  $\rightarrow$  ensures **uniform suspension**.
- > Enhances product shelf life and bioavailability in liquid form.

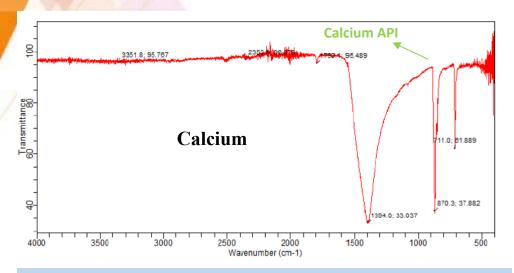


## **3b. Absorption** of Liposomal Calcium Represented Schematically on a Cellular Cross-Section





### 4a. FTIR Spectra of Calcium, Liposome & Liposomal Calcium



**Confirmation of** 2352.0; 54.648 lipid tail organization Confirmation of Hydrophilic 3852 2: 42 532 the C=O groups **Interactions** from **Phospholipids** Liposome 2853.3; 16.172 **Hydrophobic Interactions** 1399.6; 1.418 1000 Wavenumber (cm-1)

Figure 1: IR Transmission spectrum showing bands at different wavelengths of Calcium Carbonate API

Figure 2: Hydrophobic and Hydrophilic interactions within Empty Liposome

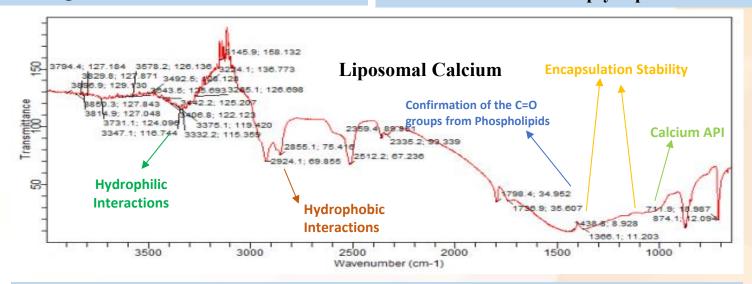


Figure 3: IR Transmission spectrum of Liposomal Calcium is shown



## 4b. Summary of FTIR Analysis of Liposomal Calcium

- 1. Confirmation of the C=O and OH groups Broad -OH peaks (~3400 cm<sup>-1</sup>) and C=O shifts (1399.6 cm<sup>-1</sup>) indicate controlled release.
- 2. Hydrophobic Interactions Distinct CH<sub>2</sub> peaks at 2924 cm<sup>-1</sup> and 2855 cm<sup>-1</sup> confirm ordered lipid tail packing with Calcium carbonate.
- 3. Hydrophilic Interactions Broad -OH peaks (~3400 cm<sup>-1</sup>) and CO<sub>3</sub><sup>2-</sup> peaks (711.9 cm<sup>-1</sup>) confirm strong hydrophilic interactions.
- 4. API Shifts in C=O stretching (~1366.1 cm<sup>-1</sup>), consistent CH<sub>2</sub> bands (~2924, 2855 cm<sup>-1</sup>) and CO<sub>3</sub><sup>2-</sup> peaks (~711.9 cm<sup>-1</sup>) confirm the integration of Calcium carbonate within the phospholipid bilayer.
- 5. Encapsulation Stability Peaks at 1366.1 cm<sup>-1</sup> (C=O stretching) and 711.9 cm<sup>-1</sup> (CO<sub>3</sub><sup>2-</sup> bending) confirm successful interaction between API and lipid bilayer.



### 5. Elemental Analysis of Liposomal Calcium

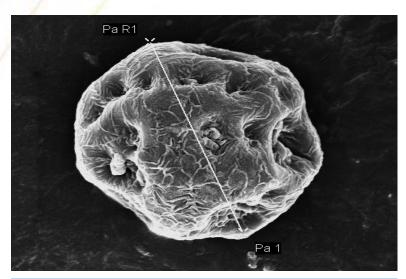
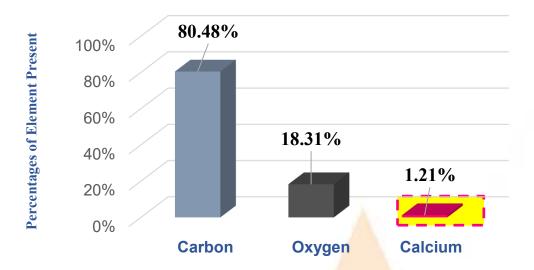


Figure 1 – SEM imaging showing the area scanned using Energy Dispersive X-Ray Spectroscopy (EDAX)

- No surface calcium detected in the liposomal calcium spectrum, confirming that calcium is fully enclosed and not exposed on the surface.
- ➤ EDAX scan shows that only the liposomal shell elements are detected, proving that the calcium core is completely encapsulated within the liposome.

#### (a) ELEMENTAL COMPOSITION OF CALCIUM API



#### (b) ELEMENTAL COMPOSITION OF LIPOSOMAL CALCIUM

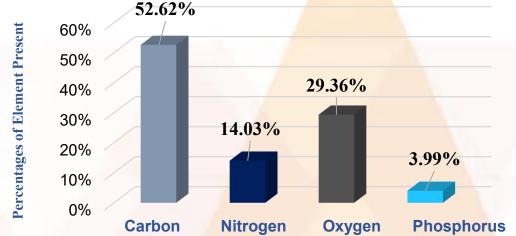
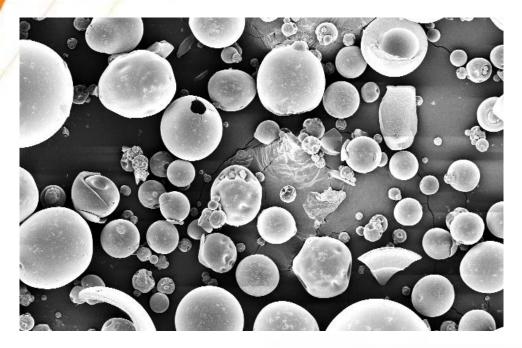


Figure 2 – A graphical representation of the percentages of elements composing (a) Calcium API and (b) Liposomal Calcium



## 6. Morphology of Liposomal Calcium as Viewed Under a Scanning Electron Microscope



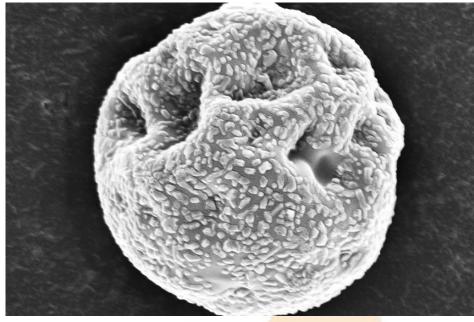


Figure 1 – SEM image of few Calcium Liposomes scattered within the field of view under observation

Figure 2 – Zoomed-in view of a Calcium Liposome

- Spherical morphology observed in liposomal calcium particles.
- Uniform size distribution seen across the field (Figure 1).
- Particles appear round surfaced.
- Spherical and uniform morphology enhances stability, encapsulation efficiency, and cellular uptake, making it ideal for liposomal drug delivery.



## 7. Leakage of Calcium from Liposomes

#### MINERAL LEAKAGE ASSAY

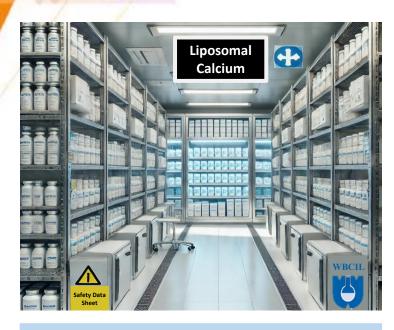


Figure 1 – An image representing the storage of formulations stored on shelves

- Encapsulation efficiency remains high (~94) throughout 3 years of storage, indicating stable liposome structure.
- > Assay values for free calcium remain low (~29%), showing minimal leakage over time.
- The formulation shows excellent retention of calcium, confirming its suitability for long-term shelf storage.

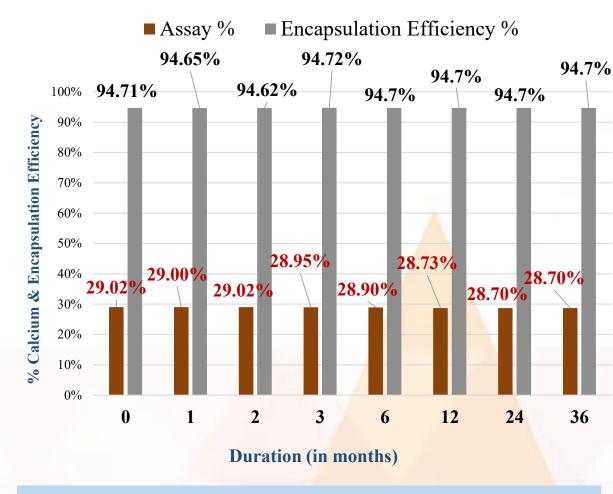


Figure 2 – Chart comparing the stability of Liposomal Calcium stored over a period of 3 years at  $40^{\circ}\text{C} \pm 2^{\circ}\text{C}$  and a relative humidity of 75%  $\pm$  5%.



## 8. Stability of Liposomal Calcium at Elevated Temperatures



Figure 1 – An image representing the transport of formulations at elevated temperatures.

- Encapsulation efficiency remains high (~94.7%) even after exposure to 105°C for 4 hours.
- Assay values (29.02% at RT vs. 28.74% at 105°C) show minimal variation, indicating negligible calcium leakage.
- ➤ Demonstrates **thermal robustness**, making the formulation suitable for transport and storage in hot climates.

#### TEMPERATURE EXPOSURE STUDY

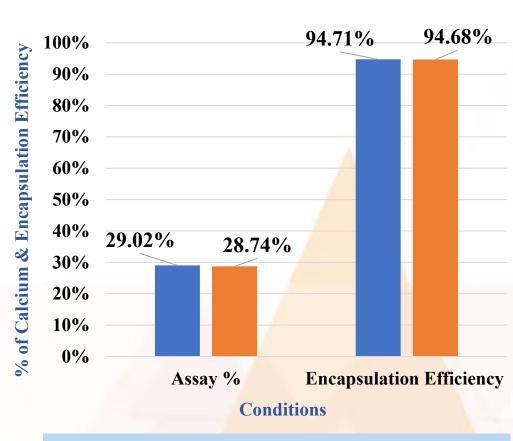
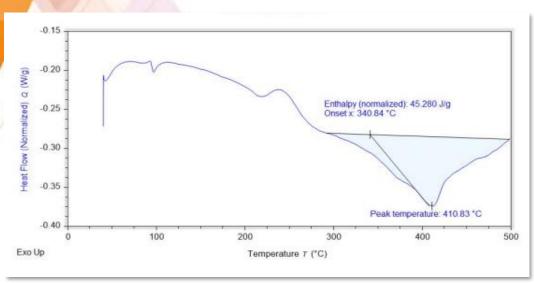


Figure 2 – Chart comparing the stability of Liposomal Calcium both at room temperature (RT) and at 105°C (exposure for 4 hours).



### 9a. Differential Scanning Calorimetry Analysis



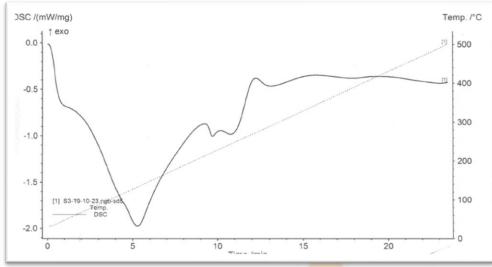


Figure 1: DSC Thermogram of Calcium API

Figure 2: DSC Thermogram of Empty Liposome

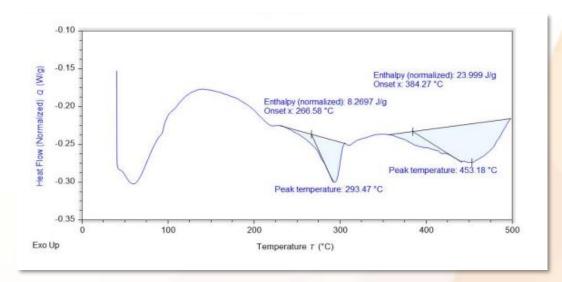
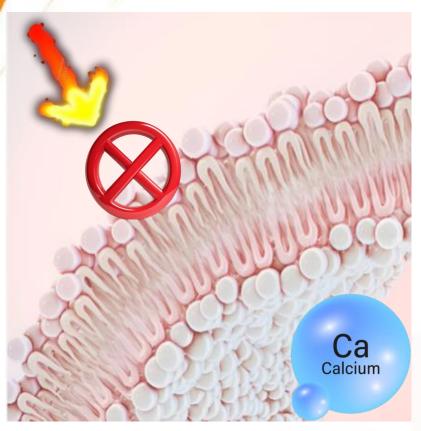


Figure 3: DSC Thermogram of Liposomal Calcium



## 9b. Endothermic study of Liposomal Calcium Using Differential Scanning Calorimetry Analysis



An illustration showing how the phospholipid bilayer is preventing the heat to accessing Calcium API from getting disintegrated.

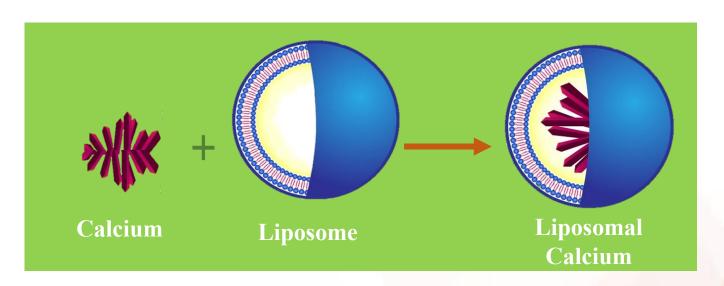
| Sample                   | Thermal<br>Events<br>(°C)    | Inference   |
|--------------------------|------------------------------|---|
| Calcium<br>Carbonate API | 410.83                       | Distinct thermal transitions indicate phase changes or melting points associated with Calcium carbonate*.           |
| Liposome                 | 136.85,<br>212.78,<br>278.42 | Exhibits multiple transitions related to phospholipid structural changes and thermal stability*.                    |
| Liposomal<br>Calcium     | 293.47,<br>453.18            | Enhanced thermal stability indicates strong molecular interaction between the lipid bilayer and calcium carbonate*. |



## 10. Mineral Loading Capacity

**Calcium Loading** 

0.76 mg per mg of Liposomal Product



Formulation of Calcium in Liposomes

- Calcium loading capacity in Liposomes refers to the amount of Calcium encapsulated within the Liposome relative to the total weight of the Liposomal formulation.
- A higher Calcium loading capacity in Liposomes ensures more efficient mineral delivery, reduces the amount of Liposome required, and improves therapeutic outcomes.



### 11. Product Uniformity - Mesh Size vs EE%

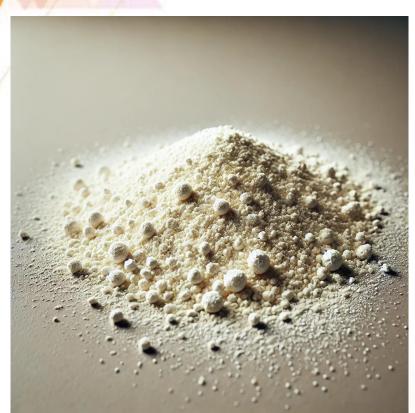


Figure 1 - A representative image of Liposomal Calcium powder.

#### GRAIN SIZE ANALYSIS USING MESH OF VARIED POROSITY

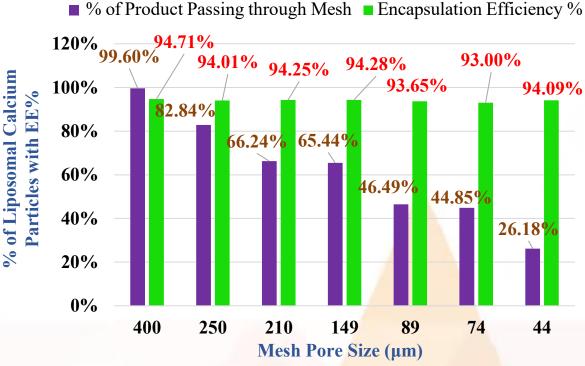


Figure 2 – Chart comparing the % of Liposomal Calcium that can pass through mesh of varied porosity with their respective encapsulation efficiency percentages.

- Encapsulation efficiency remained stable (93-94%) across all mesh sizes, regardless of particle retention.
- This indicates that calcium is effectively entrapped within liposomes of all sizes, confirming formulation integrity and uniform encapsulation.

# Thank You!!!









